2013 Vol. 15, No. 1 26–29

Gold-Catalyzed Sequential Alkyne Activation for the Synthesis of 4,6-Disubstituted Phosphorus 2-Pyrones

Juntae Mo,[†] Dongjin Kang,[†] Dahan Eom,[†] Sung Hong Kim,[‡] and Phil Ho Lee*,[†]

Department of Chemistry, Kangwon National University, Chuncheon 200-701, Republic of Korea and Analysis Research Division Daegu Center, Korea Basic Science Institute, Daegu 702-701, Republic of Korea

phlee@kangwon.ac.kr

Received October 23, 2012

ARSTRACT

Tandem gold-catalyzed addition of alkynyl phosphonic acid monoethyl esters to terminal alkynes and cyclization were developed for the synthesis of 4,6-disubstituted phosphorus 2-pyrones in one reaction vessel based on the concept of sequential alkyne activation. Alkynyl enol phosphonates were selectively obtained through the gold-catalyzed addition reaction in the presence of a catalytic amount of triethylamine. Also, gold-catalyzed cyclization of alkynyl enol phosphonates was successful in giving a variety of 4,6-disubstituted phosphorus 2-pyrones.

2-Pyrones are an important structural motif found not only in valuable biologically active compounds¹ but also in diverse synthetic intermediates.² Recently, the activity of

2-pyrones as potent HIV protease inhibitors led to further investigations of 2-pyrone and its derivatives.³

Organophosphorus compounds continue to receive widespread attention due to their ubiquity in biological systems⁴ and their potential to serve as novel pharmaceuticals.⁵ Since there is a remarkable similarity in reactivity and bioactivity between the carbon species

$$(MeO)_2PCH_2CHO + R = Me, EtO R$$

$$(1)$$

$$R = Me, EtO R$$

$$(1)$$

$$R = Me, EtO R$$

$$(1)$$

$$R = Me, EtO R$$

$$(2)$$

$$R = Me, EtO R$$

$$(3)$$

$$R = H, Br$$

$$R = H, B$$

and their phosphorus counterparts, 6 we imagined phosphorus 2-pyrones to have potential bioactivities similar to

[†] Kangwon National University.

^{*}Korea Basic Science Institute.

^{(1) (}a) Mori, T.; Ujihara, K.; Matsumoto, O.; Yanagi, K.; Matsuo, N. J. Fluorine Chem. 2007, 128, 1174. (b) McGlacken, G. P.; Fairlamb, I. J. S. Nat. Prod. Rep. 2005, 22, 369. (c) Schlingmann, G.; Milne, L.; Carter, G. T. Tetrahedron 1998, 54, 13013. (d) Shi, X.; Leal, W. S.; Liu, Z.; Schrader, E.; Meinwald, J. Tetrahedron Lett. 1995, 36, 71. (e) Barrero, A. F.; Oltra, J. E.; Herrador, M. M.; Sanchez, J. F.; Quilez, J. F.; Rojas, F. J.; Reyes, J. F. Tetrahedron 1993, 49, 141.

^{(2) (}a) Ryu, K.; Cho, Y.-S.; Jung, S.-I.; Cho, C.-G. Org. Lett. 2006, 8, 3343. (b) Fairlamb, I. J. S.; Marrison, L. R.; Dickinson, J. M.; Lu, F.-J.; Schmidt, J. P. Bioorg. Med. Chem. 2004, 12, 4285. (c) Shimizu, H.; Okamura, H.; Yamashita, N.; Iwagawab, T.; Nkatanib, M. Tetrahedron Lett. 2001, 42, 8649. (d) Okamura, H.; Shimizu, H.; Iwagawa, T.; Nakatani, M. Tetrahedron Lett. 2000, 41, 4147. (e) Posner, G. H.; Lee, J. K.; White, M. W.; Hutchings, R. H.; Dai, H.; Kachinski, J. L.; Kensler, T. W. J. Org. Chem. 1997, 62, 3299. (f) Posner, G. H.; Cho, C.-G.; Anjeh, T. E. N.; Johnson, N.; Horst, R. L.; Kobayashi, T.; Okano, T.; Tsugawa, N. J. Org. Chem. 1995, 60, 4617.

⁽³⁾ Lee, Y. S.; Kim, S. N.; Lee, Y. S.; Lee, J. Y.; Lee, C.-K.; Kim, H. S.; Park, H. Arch. Pharm. **2000**, *333*, 319.

^{(4) (}a) Melzer, M.; Chen, J. C.-H.; Heidenreich, A.; Gäb, J.; Koller, M.; Kehe, K.; Blum, M.-M. J. Am. Chem. Soc. 2009, 131, 17226. (b) Chanda, A.; Khetan, S. K.; Banerjee, D.; Ghosh, A.; Collins, T. J. J. Am. Chem. Soc. 2006, 128, 12059. (c) Seto, H.; Kuzuyama, T. Nat. Prod. Rep. 1999, 16, 589.

those of the 2-pyrones reported. However, so far, only four synthetic methods of phosphorus 2-pyrones have been described in the literature. In 1978, Razumov et al. reported the synthesis of two phosphorus 2-pyrones through intermolecular aldol condensation and sequential thermal cyclization reactions (eq 3). In the same year, Sigal et al. prepared phosphorus 2-pyrone by addition of bromine to mesityl-2-butenyl phostinate and successive dehydrobromination (eq 2).8 In 2002, Cremer et al. prepared two phosphorus 2-pyrones through a bromination-dehydrobromination sequence (4 steps) from the corresponding saturated phostone (eq 3). These methods depend on the functionalization of a preformed cyclic phostinate and phostone nucleus. Recently, Ding et al. reported an efficient procedure to synthesize 6-substituted phosphorus 2-pyrones through Ag-catalyzed cyclization of (Z)-2-alken-4-ynylphosphonic acid monoethylesters (eq 4). 10 However, to the best of our knowledge, a synthetic method for 4,6-disubstituted phosphorus 2-pyrones has never been reported thus far. Based on the concept of sequential alkyne activation, 11 we herein describe goldcatalyzed sequential intermolecular addition of 1-alkynylphosphonic acid monoethyl esters to alkynes followed by intramolecular cyclization of alkynyl enol phosphonates in one reaction vessel, leading to the formation of the 4,6disubstituted phosphorus 2-pyrones (Scheme 1).

Scheme 1. Synthesis of 4,6-Disubstituted Phosphorus 2-Pyrones

We initiated our investigation using phenylethynyl phosphonic acid monoethyl ester 2a and 5-phenyl-1-pentyne

(1a) (Table 1). 1-Alkynyl phosphonic acid monoethyl esters were readily prepared from the basic hydrolysis of diethyl 1-alkynyl phosphonates, which were synthesized by the reaction of lithium alkynides with diethyl phosphoryl chloride. ¹² AuCl and AuCl₃ in the presence of AgOTf or

Table 1. Optimization of Tandem Addition and Cyclization^a

entry	cat. (mol %)	temp (°C)	time (h)	yield $(\%)^b$
1	AuCl (5)	80	12	0
2	AuCl (5)/AgOTf (5)	40	24	0
3	$AuCl_3(5)$	80	12	0
4	$AuCl_3$ (5)/AgOTf (15)	40	24	0
5	Ph ₃ PAuCl (5)/AgOTf (5)	40	24	51
6	Ph ₃ PAuCl (5)/AgOTf (5)	60	24	$50 (13)^c$
7	$Ph_3PAuCl(5)/AgNTf_2(5)$	40	9	62
8	$Ph_{3}PAuCl(5)/AgSbF_{6}(5)$	40	16	55
9	$JohnPhos-Au^{d}(5)$	30	14	68
10	AgOTf(5)	80	12	0

^a 1a (0.4 mmol) and 2a (0.2 mmol) were used in DCE (0.8 mL).
^b Isolated yield of 3a. ^c 2-Ethoxy-6-methyl-4-phenyl-5-phenethyl-1,2-oxa-phosphorin 2-oxide (4a). ^d JohnPhos-Au: (acetonitrile)[(2-biphenyl)-di-tert-butylphosphine]gold(I) hexafluoroantimonate.

not failed to catalyze the tandem reaction (entries 1-4). Treatment of 1a and 2a with Ph3PAuCl and AgOTf (5 mol % each) as a catalyst selectively gave the cyclized tandem product 3a in 51% yield in DCE at 40 °C after 24 h (entry 5). The reaction was sensitive to the temperature. When this reaction was carried out in DCE at 60 °C for 24 h, 3a was obtained in 50% yield and, in addition, 2-ethoxy-6-methyl-5-phenethyl-4-phenyl-1,2-oxaphosphorin 2-oxide (4a) was isolated in 13% yield (entry 6). The noncoordinating counterion of the cationic gold(I) catalyst had a little effect on the product distribution. When AgNTf₂ or AgSbF₆ was used, we selectively isolated 3a in 62% and 55% yields, respectively (entries 7 and 8). The best result was obtained by using the catalyst (acetonitrile)[(2-biphenyl)-ditert-butylphosphinelgold(I) hexafluoroantimonate (5 mol %) in dichloroethane at 30 °C after 14 h, which gave rise to 3a in 68% yield through C–O and sequential C–C bond formation in one reaction vessel (entry 9). AgOTf alone failed to catalyze the tandem reaction (entry 10). To check the possibility of catalysis by a protic acid, we attempted the tandem reaction in the presence of trifluoromethanesulfonic acid (5 mol %) in dichloromethane at 30 °C. Under these conditions, the reaction did not proceed.

The scope of the tandem reaction was examined with a variety of alkynes 1 and 1-alkynyl phosphonic acid monoethyl esters 2 (Table 2). Treatment of 2a with 1-hexyne (1b) and 3-phenyl-1-propyne (1c) with a gold catalyst gave the

Org. Lett., Vol. 15, No. 1, 2013

^{(5) (}a) Ruda, G. F.; Wong, P. E.; Alibu, V. P.; Norval, S.; Read, K. D.; Barrett, M. P.; Gilbert, I. H. J. Med. Chem. 2010, 53, 6071. (b) Congiatu, C.; Brancale, A.; Mason, M. D.; Jiang, W. G.; McGuigan, C. J. Med. Chem. 2006, 49, 452. (c) Kafarski, P.; LeJczak, B. Curr. Med. Chem.: Anti-Cancer Agents 2001, 1, 301. (d) Colvin, O. M. Curr. Pharm. Des. 1999, 5, 555. (e) Mader, M. M.; Bartlett, P. A. Chem. Rev. 1997, 97, 1281. (f) Kafarski, P.; Lejczak, B. Phosphorus, Sulfur Silicon Relat. Elem. 1991, 63, 193

^{(6) (}a) Dillon, K. B.; Mathey, F.; Nixon, J. F. *Phosphorus: The Carbon Copy*; John Wiley & Sons: Chichester, 1998. (b) Quin, L. D. *A Guide to Organophosphorus Chemistry*; John Wiley & Sons: New York, 2000; Chapter 11.

⁽⁷⁾ Razumov, A. I.; Liorber, B. G.; Sokolov, M. P.; Zykova, T. V.; Salakhutdinov, R. A. Zh. Obshch. Khim. 1978, 48, 51.

⁽⁸⁾ Sigal, I.; Loew, L. J. Am. Chem. Soc. 1978, 100, 6394.

⁽⁹⁾ Polozov, A. M.; Cremer, S. E. J. Organomet. Chem. 2002, 646, 153

⁽¹⁰⁾ Peng, A.-Y.; Ding, Y.-X. Org. Lett. **2005**, 7, 3299.

^{(11) (}a) Luo, T.; Dai, M.; Zheng, S.-L.; Schreiber, S. L. Org. Lett. **2011**, 13, 2834. (b) Sperger, C. A.; Fiksdahl, A. J. Org. Chem. **2010**, 75, 4542. (c) Sperger, C.; Fiksdahl, A. Org. Lett. **2009**, 11, 2449. (d) Liu, X.-Y.; Che, C.-M. Angew. Chem., Int. Ed. **2008**, 47, 3805. (e) Zhao, J.; Hughes, C. O.; Toste, F. D. J. Am. Chem. Soc. **2006**, 128, 7436. (f) Hashmi, A. S. K.; Schwarz, L.; Choi, J.-H.; Frost, T. M. Angew. Chem., Int. Ed. **2000**, 39, 2285.

⁽¹²⁾ Lahrache, H.; Robin, S.; Rousseau, G. Tetrahedron Lett. 2005, 46, 1635.

cyclized 4,6-disubstituted phosphorus 2-pyrones 3b and 3c in 71% and 70% yields, respectively, in DCE at 30 °C within 16 h (entries 1 and 2). Reaction of 2a with enyne 1d in the presence of the gold catalyst gave the desired 3d in 62% yield albeit with an excess amount (5 equiv) of 1d (entry 3). Under the optimized reaction conditions, ethynyl phosphonic acid monoethyl ester 2b having an electrondonating 4-methylphenyl group was reacted with 1-hexyne (1b) and 3-cyclohexyl-1-propyne (1e) to afford 3e and 3f in good yields in one reaction vessel (entries 4 and 5). Certain labile functional groups commonly employed in organic synthesis such as chloro or ester moieties substituted on the alkyne were tolerated under these reaction conditions (entries 6 and 7). Phosphonic acid monoethyl ester 2c having an electron-withdrawing 4-chlorophenyl group underwent the Au-catalyzed tandem reaction with 1a and 1f, producing the phosphorus 2-pyrones 3i (56%) and 3i (50%) in slightly lower yields, presumably because the capacity of the triple bond of 2c to coordinate gold and the nucleophilicity of phosphonic acid monoethyl ester is diminished by the existing chloride group (entries 8 and 9). The chloride functionality provides a handle to introduce other groups at the para position of phenyl group at the C-4 position via transition-metal-catalyzed cross-coupling reactions. Subjecting 1-hexynylphosphonic acid monoethyl ester 2d to 1b and 1h produced the phosphorus 2-pyrones 3k and 3l in moderate to good yields (entries 10 and 11). When 4-phenyl-1-butyne (1i) and 5-phenyl-1pentyne (1a) were used, the corresponding desired phosphorus 2-pyrones 3m and 3n were obtained in 67% and 62% yields, respectively (entries 12 and 13). Exposure of 5-chloro-1-pentyne (1f) and 2d to the gold catalyst afforded 30 in 64% yield through intermolecular C-O followed by intramolecular C-C bond formation (entry 14).

Although the mechanism of the present reaction has not been fully established at the present stage, a possible reaction pathway is shown in Scheme 2. Coordination of a gold catalyst to the triple bond in alkyne 1 results in the formation of the intermediate A which, upon nucleophilic attack of the oxygen on the 1-alkynyl phosphonic acid monoethyl ester 2, is converted to a σ -gold complex **B**. Then, 1-alkynyl enol phosphonate 5 was produced after deprotonation followed by protodeauration of **B**. The triple bond in alkyne 1 rather than 2 was first activated by a gold catalyst, presumably because the capacity of the triple bond of 2 to coordinate gold is diminished by conjugation. Coordination of a gold catalyst to a triple bond in 5 results in the formation of the intermediate C which, upon nucleophilic attack of the carbon on the enol phosphonate, is cyclized leading to a σ -gold complex **D**. Subsequent deprotonation and protodeauration of D afforded the phosphorus 2-pyrone derivative 3 to release the gold catalyst back into the catalytic cycle. In addition, gold-catalyzed isomerization of 5 may occur at a higher temperature to produce the more substituted enol

Table 2. Au-Catalyzed Preparation of Phosphorus 2-Pyrones^a

cat. Au

.OFt

phosphonate E.¹³ The repeated activation of the triple bond in E by the gold catalyst followed by cyclization,

28 Org. Lett., Vol. 15, No. 1, 2013

^{(13) (}a) Lee, P. H.; Kang, D.; Choi, S.; Kim, S. *Org. Lett.* **2011**, *13*, 3470. (b) Lee, P. H.; Kim, S.; Park, A.; Chary, B. C.; Kim, S. *Angew. Chem., Int. Ed.* **2010**, *49*, 6806. (c) Chary, B. C.; Low, W. S.; Kim, S.; Kim, H.; Lee, P. H. *Chem.*—*Asian J.* **2011**, *6*, 1970.

^a Reactions were carried out with 1 (0.4 mmol), 2 (0.2 mmol), and (acetonitrile)[(2-biphenyl)-di-*tert*-butylphosphine]gold(I) hexafluoroantimonate (5 mol %) in DCE (0.8 mL) at 30 °C. ^b Isolated yield. ^c Alkyne (5 equiv) was used.

Scheme 2. Plausible Mechanism of Gold-Catalyzed Tandem Reaction

EtO
$$\stackrel{\circ}{P}$$
 Au $\stackrel{\circ}{R^2}$ $\stackrel{\circ}{R^2}$

deprotonation, and protodeauration would give 4,5, 6-trisubstituted phosphorus 2-pyrone 4, which is supported by the formation of a side product (4a) in entry 6 (Table 1).

Next, we attempted to isolate the intermediate 5 to prove the plausible mechanism of the present tandem reaction. Gratifyingly, 1-alkynyl enol phosphonates were selectively obtained with the gold catalyst in the presence of triethylamine (Scheme 3). ¹⁴ For example, treatment of 1a and 2a with JohnPhos-Au (5 mol %) in the presence of triethylamine (10 mol %) selectively gave phenylethynyl enol phosphonate 5a in 69% yield in DCE at 30 °C after 14 h without formation of the tandem product (3a). The addition of 2b to 1f proceeded smoothly to provide the desired enol phosphonate 5c in 81% yield. Likewise, hexynyl enol phosphonate 5d was synthesized in 73% yield from 5-chloro-1-pentyne (1f) and 2d.

Based on the above mechanistic considerations, gold-catalyzed cyclization of **5** was examined (Table 3). We were pleased to observe that treatment of **5c** with John-Phos-Au (5 mol %) selectively provided **3g** in 83% yield in DCE at 30 °C after 3 h without formation of 4,5,6-trisubstituted phosphorus 2-pyrone (entry 3). Similarly, **5b** and **5d** were smoothly converted to the corresponding 4,6-disubstituted phosphorus 2-pyrones (**3c** and **3o**) in excellent yields under the mild reaction conditions (DCE, 30 °C, 1–2 h) catalyzed by gold (entries 2 and 5). Isolation of alkynyl enol phosphonates (Scheme 3) and their cyclization to phosphorus 2-pyrones (Table 3) indicate that the gold-catalyzed sequential alkyne activation process is operated as shown in Scheme 2.

Scheme 3. Au-Catalyzed Addition of 1-Alkynyl Phosphonic Acid Monoethyl Esters to Alkynes^a

 a Conditions: 1 (0.4 mmol) and 2 (0.2 mmol) in the presence of (acetonitrile)[(2-biphenyl)-di-*tert*-butylphosphine]gold(I) hexafluoroantimonate (5 mol %) and Et₃N (10 mol %) were used in DCE (0.8 mL) at 30 °C.

Table 3. Au-Catalyzed Cyclization of 1-Alkynyl Enol Phosphonate⁴

entry	product	time (h)	yield (%)
1	3a	2	84
2	3c	2	86
3	3 g 3i	3	83
4	3 i	6	72
5	3o	1	94

"5 (0.2 mmol) and (acetonitrile)[(2-biphenyl)-di-*tert*-butylphosphine] gold(I) hexafluoroantimonate (5 mol %) in DCE (0.8 mL) at 30 °C.

In conclusion, we have developed the tandem gold-catalyzed addition of 1-alkynyl phosphonic acid monoethyl esters to terminal alkynes and cyclization for the synthesis of 4,6-disubstituted phosphorus 2-pyrones in one reaction vessel based on the concept of sequential alkyne activation. 1-Alkynyl enol phosphonates were selectively obtained through the gold-catalyzed addition reaction in the presence of a catalytic amount of triethylamine. Also, gold-catalyzed cyclization of alkynyl enol phosphonates was successful in giving phosphorus 2-pyrones.

Acknowledgment. We thank the NCRL (2012-0001245) program funded by the National Research Foundation of Korea. P.H.L. thanks Prof. S. Chang (KAIST) for helpful discussions.

Supporting Information Available. Experimental procedure and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

Org. Lett., Vol. 15, No. 1, 2013

⁽¹⁴⁾ Additional studies of the role of Et₃N in selectively giving 1-alkynyl enol phosphonates are currently underway. (a) Peng, A.; Ding, Y. *Synthesis* **2003**, 205. (b) Hua, R.; Tanaka, M. *Chem. Lett.* **1998**, 431.

The authors declare no competing financial interest.